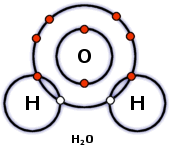
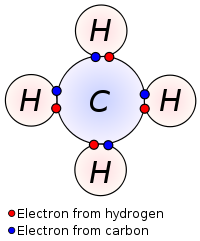
**Chemical Bonds**

* \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ bonds
* \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ bonds
* \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ forces

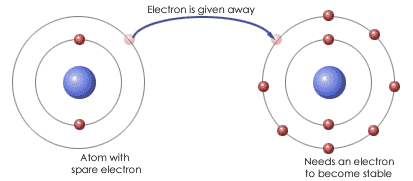
**Covalent Bonds**

Definition –



**Ionic Bonds**

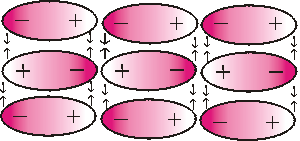
Definition –

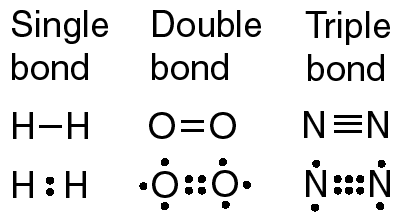


**Van der Waal Forces**

Definition –

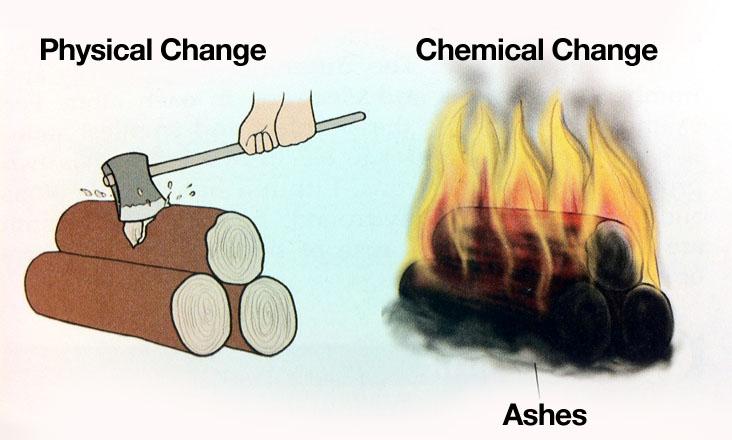
* \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ than covalent and ionic bonds
* Like a 





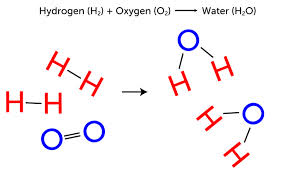
**Chemical Reactions**

* Process by which \_\_\_\_\_\_\_\_\_\_\_\_\_ reorganize into different substances
* Chemical bonds are \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ and/or formed

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**Reactant and Products**

* Reactants = \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ material
* Products= \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ material



**Parts of a Formula Equation**

